University of Groningen

Bright Ion Channels and Lipid Bilayers

Szymanski, Wiktor; Yilmaz, Duygu; Kocer, Armagan; Feringa, Ben L.

Published in:
Accounts of Chemical Research

DOI:
[10.1021/ar4000357](https://doi.org/10.1021/ar4000357)

IMPORTANT NOTE: You are advised to consult the publisher's version (publisher's PDF) if you wish to cite from it. Please check the document version below.

Document Version
Publisher's PDF, also known as Version of record

Publication date:
2013

[Link to publication in University of Groningen/UMCG research database](#)

Citation for published version (APA):
Szymanski, W., Yilmaz, D., Kocer, A., & Feringa, B. L. (2013). Bright Ion Channels and Lipid Bilayers. *Accounts of Chemical Research*, 46(12), 2910-2923. <https://doi.org/10.1021/ar4000357>

Copyright

Other than for strictly personal use, it is not permitted to download or to forward/distribute the text or part of it without the consent of the author(s) and/or copyright holder(s), unless the work is under an open content license (like Creative Commons).

The publication may also be distributed here under the terms of Article 25fa of the Dutch Copyright Act, indicated by the "Taverne" license. More information can be found on the University of Groningen website: <https://www.rug.nl/library/open-access/self-archiving-pure/taverne-amendment>.

Take-down policy

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

Downloaded from the University of Groningen/UMCG research database (Pure): <http://www.rug.nl/research/portal>. For technical reasons the number of authors shown on this cover page is limited to 10 maximum.

Bright Ion Channels and Lipid Bilayers

WIKTOR SZYMAŃSKI,^{†,§} DUYGU YILMAZ,^{‡,§} ARMAĞAN KOÇER,^{*,‡}
AND BEN L. FERİNGA^{*,†}

[†]Center for Systems Chemistry, Stratingh Institute for Chemistry and [‡]Groningen
Biomolecular Sciences and Biotechnology Institute University of Groningen,
Nijenborgh 4, 9747 AG, Groningen, The Netherlands

RECEIVED ON FEBRUARY 6, 2013

CONSPECTUS

If we look at a simple organism such as a zebrafish under a microscope, we would see many cells working in harmony. If we zoomed in, we would observe each unit performing its own tasks in a special aqueous environment isolated from the other units by a lipid bilayer approximately 5 nm thick. These confined units are social: they communicate with one another by sensing and responding to the chemical changes in their environment through receptors and ion channels. These channels control the highly specific and selective passage of ions from one side of the cell to the other and are embedded in lipid bilayers. The movement of ions through ion channels supports excitation and electrical signaling in the nervous system.

Ion channels have fascinated scientists not only because of their specificity and selectivity, but also for their functions, the serious consequences when they malfunction, and the other potential applications of these molecules. Light is a useful trigger to control and manipulate ion channels externally. With the many state-of-the-art optical technologies available, light offers a high degree of spatial and temporal control, millisecond precision, and noninvasive intervention and does not change the chemical environment of the system of interest.

In this Account, we discuss research toward the dynamic control of lipid bilayer assembly and channel function, particularly the transport across the lipid bilayer-ion channel barrier of cells using light. We first summarize the manipulation of ion channel activity with light to modulate the channel's natural activity. Based on the type of photoswitch employed, we can achieve novel functionalities with these channels, and control neural activity. Then we discuss the recent developments in light-induced transport through lipid bilayers. We focus on three different approaches: the incorporation of photoswitchable copolymers into the lipids, the doping of the lipid bilayer with photosensitive amphiphiles and the preparation of the lipid bilayers solely from photoswitchable lipids.

These examples reflect the versatility of what we can achieve by manipulating biological systems with light, from triggering the permeability of a specific area of a lipid bilayer to controlling the behavior of a whole organism.

Photochromism and Molecular Photoswitches

Numerous biological processes, including the process of vision, are enabled by interactions of matter with light. Nature has developed advanced molecules, such as rhodopsin, which undergo isomerization upon absorbing a photon of a given energy; an event that is, for example,

translated into sensual receptions. These intricate photo-responsive systems greatly stimulated the design of photochromic materials, in particular photochemically bistable molecules.

Nowadays, a plethora of synthetic molecules, called "molecular photoswitches", which change their structure and properties upon light irradiation, is available.¹ These

FIGURE 1. General types of molecular photoswitches used for the photoregulation of transport through lipid bilayers and ion channels.

FIGURE 2. The principle of engineering a mechanosensitive channel into a light-triggered ion channel. (A) Cross section of MscL channel in a lipid bilayer (yellow residue indicates cysteine 22). (B) Affinity labeling of G22C with photoswitches (star). (C) Structure of the photoswitches designed for MscL. (D) The mechanism of light-triggered activation of MscL. Left, top view of MscL from the periplasmic side of the lipid bilayer; middle, affinity labeling of G22C position in the channel pore; right, light-induced charge separation, localized buildup of charge and consequent opening of the channel.

compounds are often characterized by having two or more isomeric forms, which can be addressed in a reversible manner by illumination with light of a specific wavelength.

Molecular photoswitches have been used in the biological context,² for example, in controlling the structure and properties of peptides³ and DNA,⁴ and channel protein function.⁵ The structures and photoinduced transformations of the photoswitches most relevant for this Account are summarized in Figure 1.

Small photocaged molecules (Figure 1A) undergo irreversible bond cleavage upon irradiation, which results in the triggered exposure of bioactive functionalities.⁶ UV-irradiation of azobenzene molecules (Figure 1B) promotes the *trans*–*cis* isomerization at the central double bond, resulting in a considerable change in geometry and polarity.⁷ This process is reversible both thermally and by irradiation with visible light. In spiropyrans (Figure 1C),⁸ the C_{spiro}–O bond undergoes heterolytic cleavage upon UV-irradiation (360–370 nm) resulting in

FIGURE 3. Irreversible activation of MscL channel by a photocaged molecule, shown by patch clamp electrophysiology. Recordings were performed at +20 mV. Channel openings are shown as upward currents. (A) Schematic representation of the photocaged molecule in the engineered MscL channel. (B) Photolysis of compound **1b** liberates charged group. (C) Current flow was not observed in the dark. (D) Real-time spontaneous channel openings upon illumination (indicated by star). Adapted with permission from ref 12. Copyright 2005 AAAS.

the formation of a polar, zwitterionic conjugated system, which is accompanied by a very large change in polarity. The ring-opening can be reversed both thermally and photochemically by irradiation with visible (>460 nm) light. Malachite Green derivatives (Figure 1D) undergo heterolytic cleavage of C–CN bond, changing from a state of low polarity to a charged state of high polarity.

Transport Across Lipid Bilayers through Light-Modulated Ion Channels

Biological membranes that separate individual cells or sub-cellular functional units from their immediate external environment are composed of a mixture of lipid bilayers and proteins. While lipid bilayers do not allow the passage of polar or hydrophilic molecules across, protein-based ion channels form temporary pores in the lipid bilayer and allow the passage of specific ions after being excited by a variety of stimuli. The stimuli can be changes in the concentration of specific ligands or ions, fluctuations in temperature, changes in the membrane potential, membrane tension, or light intensity.⁹

In the past few years, several strategies were developed to re-engineer ion channels and lipid bilayers. In the first part of this Account, we will focus on light as a tool for interference with ion channel functions.

Ion Channels with Photocaged Small Molecules. One of the earliest methods for manipulating transport across lipid bilayers was targeting ion channels and receptors with a channel modulator that has been made biologically inert by a photocleavable protecting group.²

The caging technology has been applied to various neurotransmitters for controlling neuronal activity as discussed in a recent review.⁶ We explored the use of a caged light-switch to introduce light or pH-sensitivity to a mechanosensitive channel of large conductance (MscL), a membrane protein that is located in the cytoplasmic membrane of *Escherichia coli* (Figure 2A). In Nature, this channel senses the tension in the lipid bilayer upon a sudden hypoosmotic shock and functions as a safety valve to protect bacteria from lysing.¹⁰ Although the channel opens in response to tension, it can also gate spontaneously upon increase in the

FIGURE 4. MscL gating with a photocaged pH-modulator. (A) Schematic representation of the photocaged pH-modulator with the engineered MscL channel. (B) The structure of compound **3a** and removal of the protecting group by illumination. (C) Activity of the modified MscL channels in the fluorescence dequenching assay. Channels modified with caged modulator (**3b**) do not open at any pH in the dark (light gray bars). After irradiation, channels become activated (**3c**) and released the liposomal content depending on the pH (dark gray bars). If the pH of the protein sample, which was illuminated at low pH (**3d**), immediately raised to 7.7, the release of dye stops (bar with gray gradient), due to closing of the channel; if the pH is kept constant at low pH, the channel stays open (white bar) and release continues. Adapted with permission from ref 13. Copyright 2006 Wiley-VCH Verlag GmbH & Co. KGaA.

hydrophilicity of its pore region (Figure 2B–D).¹¹ This channel has been proposed by us as a remote-controlled release valve in liposomal drug delivery devices and sensory applications.¹²

We designed synthetic compounds that undergo light-induced charge separation, localized buildup of charge, and opening of the channel (Figure 3A). The caged compound **1a** (Figure 3B) comprises a cysteine-selective alkylating agent bearing a photocleavable protecting group. After its specific binding to the pore restriction of the engineered MscL channel, in which the 22nd amino acid is converted into a cysteine (G22C), the channel proteins (**1b**, Figure 3B) were reconstituted into artificial lipid bilayers and the ionic transport through a single channel was recorded by patch clamp measurements.¹² There was no current flow observed through the channel in the dark (Figure 3C). However, when the proteoliposomes were exposed to UV light (>300 nm), the photolysis of the protecting group **1b** liberated charged acetate **2** in the channel pore, which in return opened the channel spontaneously (Figure 3D).

Furthermore, we managed to render this channel pH-sensitive on command¹³ (Figure 4A). To this end, the amino group of a pH-modulator is coupled to a photolabile group (Figure 1A) that can be removed by irradiation with UV light (366 nm) (Figure 4B). The functionality of the channels was tested in a fluorescence dequenching assay using synthetic liposomes with embedded MscL channels. In the dark, the channel modified with **3a** (**3b**, Figure 4B) remains closed, irrespective of the environmental pH (Figure 4C, light gray bars). After removal of the protecting group at the amine moiety by UV irradiation (**3c**, Figure 4B), the channel becomes responsive to pH, which results in high release of the liposomal content at low pH due to the protonation of the amine group (**3d**) (Figure 4C, white bars).

Spiropyran-Modified Ion Channels. The light-induced conversion from the spiropyran to merocyanine form (Figure 1C) also causes a dramatic increase in surface wettability, that is, increase in the hydrophilicity of the pore lining.¹⁴ We applied this reversible change in wettability in the MscL mechanosensitive channel as a mean of reversible

FIGURE 5. Light-induced reversible activation of MscL shown by patch clamp electrophysiology. Recordings were performed at +20 mV. (A) The sequential illumination of the patch with UV and visible light results in an activated and deactivated channel, respectively. (B) Structure of compound **4** used for the modification of MscL. (C) Enlarged view of the channel openings at the end of each stimulation. Adapted with permission from ref 12. Copyright 2005 AAAS.

control over its gating in the absence of its native trigger, that is, tension.¹² Compound **4a**, based on a spiropyran core attached to a cysteine-selective iodoacetate moiety, was coupled to the amino acid 22 at the hydrophobic constriction of MscL to give **4b** (Figure 5B). Upon irradiation at 366 nm, photochemical ring-opening takes place, resulting in a polar form and higher wettability allowing the passage of water and ions through the pore (**4c**, Figure 5B). Exposure to visible light (>460 nm) results in the reverse, ring-closing reaction, restoring the original, nonpolar state (**4b**, Figure 5B).

The modified channel was reconstituted into artificial lipid bilayers and its activity was monitored by patch clamp measurements. The channels conducted ions only after illumination with UV light. The channel closure could be obtained by exposure to visible light (Figure 5). Light-sensitive MscL channels were also tested for their ability to release a

liposomal cargo in a fluorescence dequenching assay (Figure 6A). For both MscL **1b** and MscL **4b** the liposomal content could be released upon photoactivation of the channels (**2** and **4c**, respectively), whereas in the dark there was no release (Figure 6B). This system offers noninvasive control over drug release from liposomal delivery systems.

Azobenzene-Modified Ion Channels. nAChR is an ionotropic receptor at the neuromuscular junctions and member of a class of transmembrane ion channels that are opened or closed in response to the binding of a chemical messenger. Binding of acetylcholine results in structural changes of the nAChR and opens an internal pore (Figure 7A). Entrance of Na⁺ ions through this pore causes depolarization of muscle cells. In 1971, a noncovalent photoswitchable agonist, the symmetric meta-substituted azobenzene bearing two benzylic trimethylammonium ions, was shown to activate the receptor in *Electrophorus* electroplaques.¹⁵ The *cis*- and *trans*-azobenzene isomers activated the receptor to different degrees. Subsequently, nAChR was engineered by covalently attaching compound **5** (Figure 7) to an intrinsic cysteine in nAChR. Photoswitching the azobenzene unit between the two forms reversibly activated the nAChR channel.^{16,17}

The strategy of converting a free photoswitchable ligand into a tethered one by adding a reactive group that attaches to the protein of interest has also been applied to Voltage-gated potassium channel.¹⁸ These channels play a crucial role in the nervous system of higher organisms as regulators of membrane potential during action potential propagation. Therefore, the ability to interfere with the functioning of these channels by light allows manipulating the electrical activity of neurons. Banghart and co-workers presented an azobenzene-modified photochemically controlled K⁺ channel SPARK (**S**ynthetic **P**hoto-isomerizable **A**zobenzene **R**egulated **K**⁺ channel).¹⁸ The switch molecule **6** (Figure 7) used in this study comprises an azobenzene, connected to a reactive maleimide, which allows attachment to a cysteine that is introduced into a Shaker channel. A quaternary ammonium group, which blocks the pore of K⁺ channels, was connected to the other end of the azobenzene. (Figure 7B). The system was designed so that the quaternary ammonium can reach the pore and block ion conduction when the azobenzene is in its elongated *trans* form but not in its bent *cis* form. As a consequence of UV-induced photoisomerization of the azobenzene unit from *trans* to *cis*, it unblocks the channel (Figure 7B). Irradiation with visible light restores the blocked state by accelerating the reverse *cis*-to-*trans* conversion. Due to the thermal *cis*-*trans* isomerization, the reblocking also occurs spontaneously in the dark.¹⁸

FIGURE 6. Photochemical control over cargo release through MscL channel in large unilamellar vesicles. (A) Left, closed MscL channel in the dark; right, light-induced opening of MscL channel and the release of liposomal content. (B) Controlled release through light-activated MscL-1b (left) and MscL-4b (right). Open squares indicate the release in the dark, and solid squares indicate the release through light-activated MscL. Adapted with permission from ref 12. Copyright 2005 AAAS.

The effect of compound **6** was tested on Shaker channels expressed in *Xenopus laevis* oocytes. When the photoswitch was tethered to the extracellular surface of the channels, ultraviolet light unblocked the channel, producing as much as 1 nA current, while visible light reblocked the channels almost completely. Both forward and backward isomerizations occur within 5 s under standard epifluorescence illumination (Figure 8).

This concept was extended to ionotropic glutamate receptors (iGluRs),¹⁹ which mediate the majority of the excitatory neurotransmission processes in the vertebrate central nervous system. The advantage of this system is that these channels are well characterized and their 3D structures of complexes with a number of agonists and antagonists are available.²⁰ The X-ray structure of the ligand binding domain of iGluR6, complexed with agonist (2S,4R)-4-methyl glutamate, was used in the design of a tethered agonist **7** (Figure 7C).²⁰ Covalent attachment to the clamshell-type ligand-binding domain of iGluR resulted in a light-activated channel, called “LiGluR”. Light-induced isomerization resulted in the bent *cis* configuration and directed the agonist to the binding site and activated the receptor (Figure 7C). With 500 nm light, the agonist was withdrawn and the receptor

was deactivated. When LiGluR was expressed in HEK293 cells, whole-cell patch clamp experiments showed that LiGluR was activated both by free glutamate and by illumination.¹⁹

Recent advances have enabled optogenetic studies on interference with the activity of live neurons with light, through genetically introducing light-sensitivity in ion channels present in cells.⁹ In the first azobenzene-based control of neuronal activity by light, **6**-tethered K⁺ (SPARK) channels were used for silencing neuronal activity. The channels were first modified through genetic mutations in order to render the photoswitch the primary regulator of the channel gating. Subsequently, the modified SPARK channels were genetically expressed in rat hippocampal neurons. Neurons were incubated for 15 min with compound **6** before electrical measurements. Exposure to 390 nm light silenced spontaneous action potentials within 3 s, and exposure to 500 nm light restored activity. Activity could also be restored by leaving neurons in the dark after silencing. Light-sensitive action-potential firing was observed only with the neurons that were expressing Shaker channels (Figure 9).

In an attempt at controlling a function of a whole organism with light, Scott et al. interfered with the behavior of zebrafish by using LiGluR.²¹ LiGluR was selectively expressed

FIGURE 7. Azobenzene-modified ion channels. (A) nAChR. (B) Potassium channel (SPARK). (C) iGluR. Panels (A) and (C) adapted with permission from ref 16. Copyright 2006 American Chemical Society. Panel (B) reproduced with permission from ref 18. Copyright 2004 Nature Publishing Group.

FIGURE 8. Photoswitching of SPARK channels. (A) UV light opens channels and visible light closes channels in an inside–out patch taken from a **6**-treated *Xenopus* oocyte expressing SPARK channels. (B) SPARK channels close slowly in the dark as the photoswitch relaxes back to the *trans* configuration. Adapted with permission from ref 18. Copyright 2004 Nature Publishing Group.

in specific neurons of zebrafish that mediate the touch-evoked escape response. The neurons were numbed upon long illumination at 380 nm, followed by a period of darkness, and the zebrafish did not respond to a mechanical poke. However, a strong pulse of light focused along the spine could trigger a direct escape away from the optical poke (Figure 10).

Due to its biocompatibility and spatiotemporal resolution, light has been an excellent tool to manipulate transport across lipid bilayers by interfering with the functions of special

ion channels not only at the single molecule level but also by control functioning of a whole organism. The spatiotemporal resolution intrinsic to optical technology matches well with the submicrometer and submillisecond cellular processes that underlie for instance neural activity. Genetic methods allow expressing light-sensitive proteins in cells and subcellular compartments or allows introduction of suitable anchoring groups to introduce synthetic photoswitches such as light-sensitive caged compounds, reversibly caged ligands, and photochromic

FIGURE 9. Expression of light-activated channels confers light sensitivity on hippocampal pyramidal neurons. (A) Spontaneous action potentials are silenced and revived by exposure to 390 and 500 nm light, respectively. (B) Summary of repetitive firing data. Number of spikes resulting from a suprathreshold depolarization to -15 mV is significantly modulated by light in the multiply mutated Shaker-transfected neurons treated with ligand **6** ($*P < 0.01$). Neurons expressing the channel without **6** treatment or treated with **6** without channel expression were unaffected by light. Adapted with permission from ref 18. Copyright 2004 Nature Publishing Group.

ligands with affinity labeling capabilities at a designated position in the protein. However, the use of light in lipid bilayers is not restricted to ion channels. Recently, there have been several studies toward rendering lipid bilayers themselves light-sensitive for different applications as explained below.

Light-Modulated Transport through Lipid Bilayers

Photomodulation of transport directly through the lipid membrane²² using a stimuli-responsive bilayer forms a simple alternative to the above-described approach based upon photoswitching of membrane channel function. While it does not offer the same level of selectivity with respect to cargo being transported, the generality of this approach renders it a useful tool for delivery systems.

This part of the Account focuses on recent examples of photocontrolled systems, in which the permeability of the lipid bilayer can be altered by cycles of irradiation, while the integrity of the membrane is retained, albeit often a change in its morphology is observed. Selected cases of phototriggered, that is, irreversible release systems,²³ are also provided, in which the bursting of vesicles is observed.

Three general strategies are used to establish photocontrol over the permeability of a lipid bilayer (Figure 11). The first one (Figure 11A) employs molecular photoswitches, usually in a form of copolymers, which are incorporated in

the lipid bilayer. The second approach (Figure 11B) is based on photoswitchable lipids which are used as dopants in the membrane. Alternatively, the bilayer can be constructed completely from photochromic lipids (Figure 11C).

Incorporation of Photoswitchable Copolymers into Lipid Bilayers. The first method, pioneered recently by the group of Tribet,^{24,25} is based on the change in the polarity of azobenzene-bearing copolymer **8** (Figure 12) upon light-induced *trans*–*cis* isomerization. Hydrophobically modified poly(acrylic acid) derivatives (HMPAs), such as **8**, have the ability to bind to lipid bilayers through noncovalent, hydrophobic interactions. Upon coassembly into the lipid bilayer, they form transient nanochannels, retaining the integrity of the membrane. This process strongly depends on the hydrophobicity of HMPA, which can be affected by photoisomerization of azobenzene residues incorporated in the copolymer structure (Figure 13A). In particular, the light-induced effect has been attributed to the deeper penetration of *trans*-**8** into the membrane.²⁶

This principle has been demonstrated by using **8** to control the translocation of molecules, peptides, and quantum dots through the membranes of GUVs^{24,25} and cells.²⁵ An interesting example of an application in a highly biorelevant context was provided by incubation of mammalian CHO cells with biotinylated RL-9 peptide and either *cis*-**8** (more polar) or *trans*-**8** (less polar).²⁵ After incubation, the internalized RL-9 was visualized (Figure 13B) by conjugation with streptavidin-AlexaFluor488 (green), providing the proof of membrane permeabilization by compound **1** selectively in its apolar, *trans* form.

This relatively new approach (Figure 11A) shows great promise, due to little disruption to the membrane and proven compatibility with living cells, which is a unique achievement when compared to other methods (vide infra). What remains to be seen is the generality of the method with respect to the different architectures of lipid bilayers.

Doping the Lipid Bilayer with a Photoswitchable Amphiphile. This represents the most established approach, with its roots in the 1970s, when naturally occurring rhodopsin²⁷ was used as dopant. Since then, a plethora of synthetic amphiphilic molecules was designed to play the role of photosensitizing dopant (e.g. compounds **9**–**11**, Figure 12).

Compound **9**, introduced by the group of Yianni and Morgan,²⁸ has been used extensively in a series of studies²⁹ and became a “workhorse” for this type of research. It has found applications in the controlled fusion of liposomes,³⁰ fast light-triggered solute delivery,³¹ and in stepwise, wavelength-programmed control of cargo release.³² Its action is believed to rely on the increased bulkiness of the photoisomerized, *cis*-isomer, which interferes with the bilayer packing.³⁰

FIGURE 10. Light-control over zebrafish response to poking. (A) LiGluR remains activated (for tens of minutes) after brief exposure to 380 nm light. (B) A hippocampal neuron expressing LiGluR fires action potentials when illuminated at 374 nm for 5 ms and then continues to fire in the dark until 488 nm light restores native activity. (C) Zebrafish naturally respond to a mechanical poke in the side by making a directed C-turn to escape the stimulus. (D) Illumination at 380 nm, followed by a long period of darkness induces a numbing of the escape reflex, which is then restored after exposure to 500 nm light. (E) A focused optical poke with high-intensity 380 nm light prevents light from entering the eyes and induces the same C-turn escape maneuver as in panel (C). Adapted with permission from ref 21. Copyright 2007 Elsevier Inc.

In the attempt to resolve the issue of destabilization of liposomes, which is often caused by doping with a photo-switchable amphiphile and which results in the spontaneous release of cargo,³³ the group of Wang designed and applied a cholesterol-based, photoswitchable amphiphile **10**.³⁴ The spontaneous release of calcein from liposomes, formed from egg phosphatidylcholine with 20% **10**, was studied, and it was reported that the addition of cholesterol greatly stabilizes the liposomes and helps to retain the cargo. Notably, the transport through the membrane could be reversibly photoregulated by alternating irradiation with UV and Visible light.

In a follow-up study, cholesterol derivative **11** was used for the preparation of vesicles aimed at stabilizing vitamin C (in the dark) and its release under the action of UV light, toward applications in sunscreen preparations.³⁵ This constitutes a rare and noteworthy example of commercial application-driven research in the field covered by this Account.

A distinctly different approach to the design of photosensitizing dopant was presented recently by Kimura and co-workers.³⁶ It relies on the use of lipophilic malachite green derivative **12**, which upon irradiation with UV light undergoes a heterolytic cleavage of C–CN bond (vide supra), resulting in the formation of a positively charged amphiphile. A detailed

study on the possible mechanism of UV-induced permeation of a lipid membrane doped with **12** (Figure 14A) suggests³⁶ that at low concentrations of **12** (<2%) the irradiation results in the change of the vesicle shape. Only when the concentration of **12** is above 2%, the change in interactions of the amphiphile with the bilayer result in rupturing the vesicle structure. This assumption was supported by the study on the influence of the concentration of compound **12** on the photoinduced release of cargo from the vesicles (Figure 14B).

Low effective concentrations (<3%) of dopant and important mechanistic insight render this approach a valuable

addendum to the toolbox of methods aimed at externally controlling the transport through lipid bilayers.

Doping of lipid bilayers with photoswitchable amphiphiles (Figure 11B) has reached its adolescence as a method of photocontrolling the transport through membranes.

FIGURE 11. Strategies applied in the construction of lipid bilayers with photocontrolled permeability, including the doping of a bilayers with photochromic molecules (A) and photoswitchable amphiphiles (B), and construction of bilayers solely from photoswitchable amphiphiles (C). Dark and light blue: molecular photoswitchable structures in two different, photoisomeric forms.

FIGURE 13. (A) Photocontrolled membrane permeabilization by compound **8**. Apolar, *trans*-**8** binds to the membrane via hydrophobic interactions and forms nanochannels. More polar *cis*-**1** does not cause content release. Adapted with permission from ref 24. Copyright 2010 American Chemical Society. (B) Photocontrolled internalization of a peptide RL-9 into mammalian CHO cells. Reprinted with permission from ref 25. Copyright 2012 WILEY-VCH Verlag GmbH & Co. KGaA.

FIGURE 12. Photoswitchable compounds employed in the preparation of lipid bilayers with photomodulated permeability. Photoswitchable moieties are highlighted in blue.

FIGURE 14. Dopant **12** (MGL) used for the light-triggered release of a fluorescence dye ANTS from lipid vesicles. (A) Proposed, concentration-dependent mechanisms of membrane permeation. (B) Phototriggered release of ANTS at different concentration of dopant **12**. Adapted with permission from ref 36. Copyright 2010 American Chemical Society.

Initial hurdles, such as liposome destabilization and subsequent spontaneous cargo release seem to have been overcome, and the loadings of dopant can be decreased to <5%, bringing this approach closer to applications in biological systems.

Preparation of Lipid Bilayers Solely from Photoswitchable Lipids. This approach was used for the formation of cerasomes that release their cargo from their liposomal membrane in a photocontrolled manner.³⁷ Liposomes were made from amphiphilic compound **13**, which additionally incorporates in its structure both residues responsible for photoswitching (azobenzene) and formation of a porous silicate framework (Figure 15).

The light-responsive cerasomes showed expected photochemical properties, that is, reversible change in UV–vis absorption upon alternating irradiation with 365 nm (UV) and 450 nm (vis). Importantly, the controlled release of cargo (Nile Red dye) in aqueous environment could be achieved by irradiation at 365 nm, which results in the *trans*–*cis* isomerization of the azobenzene moiety, disruption of membrane packing, and, as a consequence, the increase of the bilayer permeability (Figure 15).³⁷

This novel approach to photocontrolling membrane transport (Figure 11C), although obviously limited to artificial systems, holds promise due to robustness and generality, which validate further research into the cargo compatibility and the use in biological environment.

In the context of hybrid, phototriggered delivery systems, we have recently reported³⁸ the formation of rigid nanotubes, end-capped with lipid vesicles (Figure 16). Amphiphile **14** (Figure 16A), used for the construction of the nanotubes, incorporates in its structure a photoactive moiety which

FIGURE 15. Morphology of cerasomes exhibiting photocontrolled cargo release. Adapted with permission from ref 37. Copyright 2011 Royal Society of Chemistry.

undergoes isomerization and cyclization upon irradiation with UV light. Compound **14** is able to self-assemble, probably via interdigitation of aliphatic chains (Figure 16B), forming 3 nm thick and highly stable bilayers that close into well-defined nanotubes. Importantly, when phospholipid 1,2-dioleoyl-*sn*-glycero-3-phosphate (DOPC) is added, phase separation between the amphiphile and DOPC results in a formation of unique nanotubes end-capped with lipid vesicles (Figure 16C). These end-caps can be removed and attached without affecting the nanotubes (Figure 16D).

Intiguing possibilities, which put our research in the context of smart delivery systems, are offered by the fact that the nanotubes can be disassembled photochemically. Irradiation with UV light causes cyclization of **14** followed by bursting and subsequent disassembly of the tubes, leading to small vesicles. Not only does this process take the advantage of the high spatiotemporal resolution using light, but its kinetics (rate of disassembly) can also be regulated by changing the wavelength and intensity of irradiation. It was

FIGURE 16. Hybrid, phototriggered system built from nanotubes end-capped with lipid vesicles. (A) Structure of compound **14**, the building block for nanotubes, prior (left panel) and after (right panel) irradiation with UV light. (B) Compound **14** interdigitates, forming 3 nm thick bilayer. (C) Bilayers close to form nanotubes, which can be end-capped with DOPC-based vesicles. (D) The vesicles can be removed and reattached without destroying the nanotubes; irradiation of the end-capped nanotubes with UV light results in isomerization of **14** and the disassembly of the structure. (E) Degradation of nanotubes observed in real time with fluorescent microscopy. (F) Degradation of tubes upon 365 nm light irradiation observed in real time showing formation of small, vesicle-like structures. Adapted with permission from ref 38. Copyright 2011 Macmillan Publishers Limited.

shown that the disassembly is complete in 25 s when the tubes are irradiated with higher intensity, 390 nm light (Figure 16E). On the other hand, irradiation with lower intensity, 365 nm light has to be carried out for 2 min before the tubes are degraded (Figure 16F).

Concluding Remarks

The precise control of transport across biological lipid membranes is a key event in sensing and cellular communication. With recent developments in optics, genetic, and molecular design, it has become possible to interfere in a noninvasive

manner with these transport processes by light with high spatial and temporal control.

Taking advantage of protein engineering techniques, synthetic photoswitches can be incorporated in, for example, channel proteins with high precision, allowing external control of channel function in vivo. Recently, fascinating opportunities arise from related in vivo approaches controlling neural functions of organisms by light. Alternatively, light-responsive amphiphiles allow direct control of bilayer organization and membrane permeability.

Based on these systems, novel approaches toward sensing, transport, and delivery systems emerge. In the broader context of optogenetics and synthetic biology and with the prospect of applications ranging from novel submicroscale delivery devices to external control of neuronal activity, we can look forward to a bright and exciting future for photoactive biohybrid systems.

We would like to thank to members of our groups, past and present, who contributed significantly to the research discussed here. Financial support from the European Research Council (ERC Advanced Grant 227897 to B.L.F.; ERC Starting Grant 208814 to A.K.; and ERC-Proof of Concept Grant to A.K.) and The Netherlands Organization for Scientific Research (NWO-CW, Spinoza Grant to B.L.F. and NWO-Vidi Grant 700.57.427 to A.K.) is gratefully acknowledged. The research has received funding from the Ministry of Education, Culture and Science (Gravity program 024.001.035).

BIOGRAPHICAL INFORMATION

Wiktor Szymański received his Ph.D. degree from The Warsaw University of Technology, Poland, in 2008. Since then, he is working with Prof. Ben L. Feringa and Prof. Dick B. Janssen at the University of Groningen on the use of biotransformations in organic chemistry and the construction of photoactive bioconjugates.

Duygu Yılmaz obtained her M.Sc. degree in 2010 at the Middle East Technical University, Turkey. She is doing her Ph.D. under the supervision of Dr. Armağan Kocer at the University of Groningen. Her research interests include re-engineering of ion channels with different modulators and their applications in biological systems.

Armağan Koçer obtained her Ph.D. degree in 2000 at the Middle East Technical University, Turkey. After working as a research Scientist at BiOMaDe Technology Foundation for 6 years, she started her own research group at the University of Groningen in September 2007. Her research interests include ion channels, mechanosensation at the molecular level, biohybrid sensors, liposomal drug delivery, and synthetic biology.

Ben L. Feringa obtained his Ph.D. degree in 1978 at the University of Groningen. After working as a research scientist at Shell, he was appointed full professor at the University of Groningen in 1988 and named the distinguished Jacobus H. van't Hoff Professor of Molecular

Sciences in 2004. His research interests include stereochemistry, organic synthesis, asymmetric catalysis, molecular switches and motors, self-assembly, and nanosystems.

FOOTNOTES

*To whom correspondence should be addressed. E-mail: a.kocer@rug.nl (A.K.); b.l.feringa@rug.nl (B.L.F.).

The authors declare no competing financial interest.

[§]W.S. and D.Y. contributed equally to the preparation of the manuscript.

REFERENCES

- 1 Feringa, B. L.; Browne, W. R., Eds. *Molecular Switches*; John Wiley & Sons, Ltd.: Weinheim, 2011.
- 2 Mayer, G.; Heckel, A. Biologically Active Molecules with a "Light Switch". *Angew. Chem., Int. Ed.* **2006**, *45*, 4900–4921.
- 3 Hopmann, C.; Schmieder, P.; Domaing, P.; Vogelreiter, G.; Eichhorst, J.; Wiesner, B.; Morano, I.; Rück-Braun, K.; Beyermann, M. Photocontrol of Contracting Muscle Fibers. *Angew. Chem., Int. Ed.* **2011**, *50*, 7699–7702.
- 4 Asanuma, H.; Liang, X.; Nishioka, H.; Matsunaga, D.; Liu, M.; Komiyama, M. Synthesis of Azobenzene-Tethered DNA for Reversible Photo-Regulation of DNA Functions: Hybridization and Transcription. *Nat. Protoc.* **2007**, *2*, 203–212.
- 5 Bonardi, F.; London, G.; Nouwen, N.; Feringa, B. L.; Driessen, A. J. M. Light-Induced Control of Protein Translocation by the SecYEG Complex. *Angew. Chem., Int. Ed.* **2010**, *49*, 7234–7238.
- 6 Ellis-Davies, G. C. R. Caged Compounds: Photorelease Technology for Control of Cellular Chemistry and Physiology. *Nat. Methods* **2007**, *4*, 619–628.
- 7 Beharry, A. A.; Woolley, G. A. Azobenzene Photoswitches for Biomolecules. *Chem. Soc. Rev.* **2011**, *40*, 4422–4437.
- 8 Berkovic, G.; Krongauz, V.; Weiss, V. Spiropyran and Spirooxazines for Memories and Switches. *Chem. Rev.* **2000**, *100*, 1741–1753.
- 9 Szobota, S.; Isacoff, E. Y. Optical Control of Neuronal Activity. *Ann. Rev. Biophys.* **2010**, *39*, 329–348.
- 10 Sukharev, S.; Anishkin, A. Mechanosensitive Channels: What Can We Learn From 'Simple' Model Systems? *Trends Neurosci.* **2004**, *27*, 345–351.
- 11 Yoshimura, K.; Batiza, A.; Schroeder, M.; Blount, P.; Kung, C. Hydrophilicity of a Single Residue Within MscL Correlates with Increased Channel Mechanosensitivity. *Biophys. J.* **1999**, *77*, 1960–1972.
- 12 Kocer, A.; Walko, M.; Meijberg, W.; Feringa, B. L. A light-Actuated Nanosensor Derived From a Channel Protein. *Science* **2005**, *309*, 755–758.
- 13 Kocer, A.; Walko, M.; Bulten, E.; Halza, E.; Feringa, B. L.; Meijberg, W. Rationally Designed Chemical Modulators Convert a Bacterial Channel Protein Into a pH-Sensory Valve. *Angew. Chem., Int. Ed.* **2006**, *45*, 3126–3130.
- 14 Nicoletta, F. P.; Cupelli, D.; Formoso, P.; De Filipo, G.; Colella, V.; Gugliuzza, A. Light Responsive Polymer Membranes: A Review. *Membranes* **2012**, *2*, 134–97.
- 15 Bartels, E.; Wassermann, N. H.; Erlanger, B. F. Photochromic Activators of the Acetylcholine Receptor. *Proc. Natl. Acad. Sci. U.S.A.* **1971**, *68*, 1820–1823.
- 16 Banghart, M. R.; Volgraf, M.; Trauner, D. Engineering Light-Gated Ion Channels. *Biochemistry* **2006**, *45*, 15129–15141.
- 17 Lester, H. A.; Krouse, M. E.; Nass, M. M.; Wassermann, N. H.; Erlanger, B. F. A Covalently Bound Photoisomerizable Agonist — Comparison with Reversibly Bound Agonists at Electrophorus Electropores. *J. Gen. Physiol.* **1980**, *75*, 207–232.
- 18 Banghart, M.; Borges, K.; Isacoff, E.; Trauner, D.; Kramer, R. H. Light-Activated Ion Channels for Remote Control of Neuronal Firing. *Nat. Neurosci.* **2004**, *7*, 1381–1386.
- 19 Volgraf, M.; Gorostiza, P.; Numano, R.; Kramer, R. H.; Isacoff, E. Y.; Trauner, D. Allosteric Control of an Ionotropic Glutamate Receptor with an Optical Switch. *Nat. Chem. Biol.* **2005**, *2*, 47–52.
- 20 Mayer, M. L. Crystal structures of the GluR5 and GluR6 ligand binding cores: Molecular Mechanisms Underlying Kainate Receptor Selectivity. *Neuron* **2005**, *45*, 539–552.
- 21 Scott, E. K.; Mason, L.; Arrenberg, A. B.; Ziv, L.; Gosse, N. J.; Xiao, T.; Chi, N. C.; Asakawa, K.; Kawakami, K.; Baier, H. Targeting Neural Circuitry in Zebrafish Using GAL4 Enhancer Trapping. *Nat. Methods* **2007**, *4*, 323–326.
- 22 Wandera, D.; Wickramasinghe, S. R.; Husson, S. M. Stimuli-Responsive Membranes. *J. Membr. Sci.* **2010**, *357*, 6–35.
- 23 Wang, J.; Wu, Q.; Li, J.; Ren, Q.; Wang, Y.; Liu, X. Photo-Sensitive Liposomes: Chemistry and Application in Drug Delivery. *Mini-Rev. Med. Chem.* **2010**, *10*, 172–181.
- 24 Sebai, S. C.; Cribier, S.; Karimi, A.; Massotte, D.; Tribet, C. Permeabilization of Lipid Membranes and Cells by a Light-Responsive Copolymer. *Langmuir* **2010**, *26*, 14135–14141.

- 25 Sebai, S. C.; Milioni, D.; Walrant, A.; Alves, I. D.; Sagan, S.; Huin, C.; Auvray, L.; Massotte, D.; Cribier, S.; Tribet, C. Photocontrol of the Translocation of Molecules, Peptides, and Quantum Dots through Cell and Lipid Membranes Doped with Azobenzene Copolymers. *Angew. Chem., Int. Ed.* **2012**, *51*, 2132–2136.
- 26 Ishii, K.; Hamada, T.; Hatakeyama, M.; Sugimoto, R.; Nagasaki, T.; Takagi, M. Reversible Control of Exo- and Endo-Budding Transitions in a Photosensitive Lipid Membrane. *ChemBioChem* **2009**, *10*, 251–256.
- 27 Darszon, A.; Montal, M.; Zarco, J. Light Increases Ion and Non-Electrolyte Permeability of Rhodopsin-Phospholipid Vesicles. *Biochem. Biophys. Res. Commun.* **1977**, *76*, 820–827.
- 28 Morgan, C. G.; Sandhu, S. S.; Yianni, Y. P.; Dodd, N. J. F. The Phase-Behavior of Dispersions of Bis-Azo Pc - Photoregulation of Bilayer Dynamics Via Lipid Photochromism. *Biochim. Biophys. Acta* **1987**, *903*, 495–503.
- 29 Bisby, R. H.; Mead, C.; Morgan, C. G. Active Uptake of Drugs into Photosensitive Liposomes and Rapid Release on UV Photolysis. *Photochem. Photobiol.* **2000**, *72*, 57–61.
- 30 Morgan, C. G.; Yianni, Y. P.; Sandhu, S. S.; Mitchell, A. C. Liposome Fusion and Lipid Exchange on Ultraviolet-Irradiation of Liposomes Containing a Photochromic Phospholipid. *Photochem. Photobiol.* **1995**, *62*, 24–29.
- 31 Bisby, R. H.; Mead, C.; Mitchell, A. C.; Morgan, C. G. Fast Laser-Induced Solute Release from Liposomes Sensitized with Photochromic Lipid: Effects of Temperature, Lipid Host, and Sensitizer Concentration. *Biochem. Biophys. Res. Commun.* **1999**, *262*, 406–410.
- 32 Bisby, R. H.; Mead, C.; Morgan, C. G. Wavelength-Programmed Solute Release from Photosensitive Liposomes. *Biochem. Biophys. Res. Commun.* **2000**, *276*, 169–173.
- 33 Shimomura, M.; Kunitake, T. Fluorescence and Photoisomerization of Azobenzene-Containing Bilayer-Membranes. *J. Am. Chem. Soc.* **1987**, *109*, 5175–5183.
- 34 Liu, X. M.; Yang, B.; Wang, Y. L.; Wang, J. Y. Photoisomerisable Cholesterol Derivatives as Photo-trigger of Liposomes: Effect of Lipid Polarity, Temperature, Incorporation Ratio, and Cholesterol. *Biochim. Biophys. Acta, Biomembr.* **2005**, *1720*, 28–34.
- 35 Dai, Y. Q.; Qin, G.; Geng, S. Y.; Yang, B.; Xu, Q.; Wang, J. Y. Photo-Responsive Release of Ascorbic Acid and Catalase in CDBA-Liposome for Commercial Application as a Sunscreen Cosmetic. *RSC Adv.* **2012**, *2*, 3340–3346.
- 36 Uda, R. M.; Hiraishi, E.; Ohnishi, R.; Nakahara, Y.; Kimura, K. Morphological Changes in Vesicles and Release of an Encapsulated Compound Triggered by a Photoresponsive Malachite Green Leuconitrile Derivative. *Langmuir* **2010**, *26*, 5444–5450.
- 37 Liang, X. L.; Yue, X. L.; Dai, Z. F.; Kikuchi, J. Photoresponsive Liposomal Nanohybrid Cerasomes. *Chem. Commun.* **2011**, *47*, 4751–4753.
- 38 Coleman, A. C.; Beierle, J. M.; Stuart, M. C. A.; Macia, B.; Caroli, G.; Mika, J. T.; van Dijken, D. J.; Chen, J. W.; Browne, W. R.; Feringa, B. L. Light-Induced Disassembly of Self-Assembled Vesicle-Capped Nanotubes Observed in Real Time. *Nat. Nanotechnol.* **2011**, *6*, 547–552.