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Light-harvesting Complexes (LHCs) Cluster Spontaneously in Membrane Environment Leading to Shortening of Their Excited State Lifetimes

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The light reactions of photosynthesis, which include light-harvesting and charge separation, take place in the amphiphilic environment of the thylakoid membrane. The light-harvesting complex II (LHCII) is the main responsible for light absorption in plants and green algae and is involved in photoprotective mechanisms that regulate the amount of excited states in the membrane. The dual function of LHCII has been extensively studied in detergent micelles, but recent results have indicated that the properties of this complex differ in a lipid environment. In this work we checked these suggestions by studying LHCII in liposomes. By combining bulk and single molecule measurements, we monitored the fluorescence characteristics of liposomes containing single complexes up to densely packed proteoliposomes. We show that the natural lipid environment per se does not alter the properties of LHCII, which for single complexes remain very similar to that in detergent. However, we show that LHCII has the strong tendency to cluster in the membrane and that protein interactions and the extent of crowding modulate the lifetimes of the excited state in the membrane.

Finally, the presence of LHCII monomers at low concentrations of complexes per lipidosome is discussed.

Photosynthetic organisms evolved the capacity to harvest the energy of solar radiation and store it into chemical compounds. In vascular plants and green algae, sunlight is absorbed by a series of membrane proteins called light-harvesting complexes (LHC).3 The most abundant of these pigment-protein complexes is LHCII. The LHCs have a dual function; in low light conditions they absorb solar energy and efficiently transfer the excitation energy to the reaction center, and in high light they additionally play a role in photoprotection by dissipating the energy absorbed in excess as heat (1, 2). This last process called non-photochemical quenching (NPQ) leads to a decrease of the excited state lifetime of chlorophyll a (Chl a), limiting the possibility of Chl triplet formation and thus the production of singlet oxygen (3). The fast, on the timescale of seconds, and fully reversible part of NPQ is called qE. This mechanism is triggered by the acidification of the luminal space of the thylakoids, which activates PsbS (4) and LhcSR (5), the proteins responsible for NPQ in plants and green algae, respectively, and the violoxanthin de-epoxidase, which converts violaxanthin into zeaxanthin (for reviews, see Refs. 6 and 7). Although the precise molecular mechanism of quenching has not been fully elucidated yet, a prominent idea discussed in literature is that NPQ is regulated via conformational changes of LHCII. Those changes could be correlated to, or even caused by, LHCII aggregation (8, 9). It was observed that low pH and zeaxanthin, both occurring in high light, enhance LHCII aggregation (10). Aggregation of LHCII in vitro is accompanied by a decrease of the Chl a fluorescence yield, indicating an increased rate of energy dissipation (9, 11). The generation of new quenching sites is assumed to occur either via pigment-pigment interactions at the protein-protein interface or within the protein as a result of conformational changes (6, 12). Although the effect of aggregation on fluorescence quenching in vitro is clearly demonstrated, its role in NPQ in vivo is still under debate (13, 14).

The study of NPQ in vivo is complicated by the presence of many parallel processes that make it difficult to extract its molecular details. A large part of the studies on NPQ has then been performed in vitro on isolated complexes in detergent micelles. However, these preparations represent an over-simplified system in a non-native environment. To overcome these problems, several approaches have been put forward: (i) reducing the number of components involved in NPQ in the thylakoid membrane by using mutants lacking photosystems I and II core complexes or blocking the chloroplast translation that leads to membranes enriched in LHCII (15, 16) and (ii) studying isolated LHCII in detergent-free environment using nanodisks (17), styrene maleic acid (18), amphipoles (19), or liposomes (20–22). These methods have revealed that the lifetime of LHCII in the membrane differs substantially from its lifetime in detergent micelles, suggesting that in the natural environment the antennas assume a different and more quenched conformation.

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1 The abbreviations used are: LHCII, light harvesting complex II; NPQ, non-photochemical quenching; Chl a, chlorophyll a; Chl b, chlorophyll b; Tricine, N-[2-hydroxy-1,1-bis(hydroxymethyl)ethyl]glycine; Qy, light absorption due to electronic transition along the y-axis of the chlorophyll molecule.

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The advantage provided by liposomes in the study of light-harvesting is their flexibility; they permit the analysis of the characteristic of the antennas by varying lipid composition (21) and protein/lipid ratio (20). They make it possible to change in the composition of the membrane in a controlled way and to study the effect of the presence of different pigments or proteins on the properties of the complexes (23, 24). These characteristics make proteoliposomes an attractive system to investigate in detail the mechanism of NPQ in a lipid bilayer environment. Liposomes containing zeaxanthin and PsbS, which are two essential factors in NPQ, in addition to LHCII were described (24, 25), opening way to study the molecular mechanism of NPQ in vitro. However, to be able to disentangle the effect of lateral aggregation, conformational switches, and protein-lipid interactions, all of which have been suggested to play a role in NPQ, it is essential to know how the lipid environment influences the properties of the complex and to have full control of their organization in the proteoliposome.

In this work we investigated in detail the relation between protein crowding and the fluorescence decay, which is the reporter for the quenched state of the complexes, by generating a series of proteoliposomes with different lipid/Chl ratios. Using a combination of ensemble and single molecule spectroscopy measurements we show that the different properties observed in LHCII in the membrane, as compared with detergent micelles, are not due to a different conformation of single complexes caused by the lipid environment but due to the clustering of LHCII. The latter most likely leads to conformational changes due to protein-protein interactions. The similarity with the results in cells suggests that this is what happens in vivo.

Results

Incorporation of Chlamydomonas reinhardtii LHCII into Liposome Vesicles—To mimic the thylakoid membrane of C. reinhardtii, we generated liposome vesicles composed of an identical mix of lipids: monogalactosyl diacylglycerol (41.2%), digalactosyl diacylglycerol (26.7%), sulfoquinovosyl diacylglycerols (15.6%), and the phospholipids (11.5%) and phosphatidylcholine (5%) (26). Antenna complexes (LHCII) isolated from the membranes (Fig. 1) or reconstituted in vitro with pigments (Fig. 2) were incorporated into liposome vesicles by the removal of detergent via dialysis. The initial preparation consists of a lipid/protein molar ratio of ~230:1. The resulting proteoliposomes show absorption maxima at 672 nm and 436 nm with shoulders at 651 nm and 473 nm. The absorption spectrum is

![Graph A](Image of graph A)

**A** Absorption spectra of isolated LHCII solubilized in detergent and incorporated into lipid vesicles. The spectra are normalized to the maximum in the Qy region.

**B** Fluorescence emission spectra normalized to the maximum.

**C** Fluorescence excitation spectra recorded at very low concentrations (when 1-Transmission is equal to the absorption) compared with the absorption spectrum normalized to 1.

**D** CD spectra normalized to the negative signal at 679 nm.

![Graph B](Image of graph B)

**Graph B**

**Graph C**

**Graph D**

**Graph D**
shown in Fig. 1A where it is compared with that of detergent-isolated LHCII trimers. In general, no large differences were observed between the spectra, indicating that the incorporation into liposomes does not lead to loss of pigments. These features are in agreement with previously published works on LHCII incorporated into liposomes (21, 25, 27).

The fluorescence emission spectrum of the LHCII-containing liposomes at room temperature (Fig. 1B) consists of a single peak with a maximum at 680 nm upon excitation of Chl $a$ (440 nm), Chl $b$ (475 nm), and carotenoids (500 nm). This indicates that the excitation energy is efficiently transferred from Chl $b$ and carotenoids to Chl $a$, which confirms the correct folding of the complexes. The high efficiency of excitation energy transfer is further supported by the fluorescence excitation spectrum (Fig. 1C).

The CD spectrum depicted in Fig. 1D confirms that LHCII in liposomes and detergent have the same pigment organization. The spectrum clearly shows the typical $- + -$ signal in the Qy absorption region and the negative CD feature in the blue region associated with Chl-Chl and Chl-carotenoid interactions (28).

**Aggregation of LHCII in Liposomes**—It is known that in vitro aggregation of LHCII is accompanied by a decrease of the fluorescence quantum yield (9, 23). This phenomenon, which is induced in vitro by the removal of detergent, may also occur during or after the incorporation of the complexes into liposomes, complicating the interpretation of the results. We note that the topology of aggregation might be different in solution and in the plane of a lipid bilayer, where mainly lateral aggregates are expected. However, the topology of insertion of the complexes in the liposomes is undetermined, and we cannot exclude the presence of interactions between complexes with opposite orientation. The formation of lateral aggregates in liposomes (at a lipid/protein molar ratio of $\sim230:1$) is confirmed by the fluorescence emission spectra at 77 K (Fig. 3A), which shows a peak at 696 nm typical of aggregated antennas (29). Aggregation, quantified by the relative intensity of the 696 nm band, decreased by reducing the amount of LHCII per liposome (Fig. 3B). By recording the fluorescence decay kinetics of LHCII at different levels of aggregation, we could also confirm the tight relation between lateral aggregation and fluorescence quenching. Fig. 3B clearly illustrates that the peak at 696 nm increased with the increase of the amount of chlorophyll per liposome. At the same time, the fluorescence lifetime decreased down to a minimum of 0.97 ns (Table 1). On the other hand, the lifetime of the sample in liposomes with a high lipid/protein molar ratio of $\sim7000:1$ (0.1 $\mu$g of Chl $a$) had a similar value compared with isolated monomeric LHCII in detergent (Table 1) (30, 31). A further decrease in the amount of protein per liposome (0.05 $\mu$g of Chl $a$ equal to a lipid/protein molar ratio of $\sim14,000:1$) induced the appearance of free Chls in the sample. This is indicated by the presence of a shoulder at 650 nm in the fluorescence emission spectrum upon selective Chl $b$ excitation (Fig. 3B) and by a long lifetime component above 4 ns in the time-resolved data, typical of free chlorophylls (Table 1).

**Electron Microscopy (EM) and Single Molecule Spectroscopy**—The structure and size of liposomes were investigated by EM. In Fig. 4 EM images of proteoliposomes containing LHCII (Fig. 4A) and recombinant LHCBM1 (Fig. 4B) showed a homogeneous morphology and size distribution. The average diameter of the proteoliposomes was $\sim50$ nm, and no indication of multi-lamellar structures was observed.

To acquire more quantitative information on the composition of proteoliposomes and hence on the level of protein aggregation, we performed single molecule confocal fluorescence experiments and analyzed the fluorescence intensity, the fluorescence spectrum, and the fluorescence lifetime of single immobilized proteoliposomes. The information about the absolute fluorescence intensity together with the average fluorescence lifetime allowed us to approximate the absorption cross-section $\sigma$ of the fluorescence particle within the confocal excitation (see “Experimental Procedures”). This effectively provides the size of the cluster via the number of absorbing pigments. By dividing the absorption cross-section of a single fluorescent particle (i.e. a single proteoliposome) by the absorption cross-section of a LHCII trimer at 633 nm ($\sigma_{\text{LHCCI}} = 1.4 \times 10^{-15}$ cm$^2$) we can estimate the number of incorporated

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**FIGURE 2. Spectroscopic characterization of reconstituted monomeric LHCBM1 incorporated into liposomes.** A, absorption (in detergent and in liposome) and fluorescence excitation spectra of the recombinant protein. The spectra are normalized to the maximum in the Qy region. B, fluorescence emission spectra normalized to the maximum. a.u., arbitrary units.
A surprising outcome was the low absolute fluorescence intensity of single complexes (at 0.1 µg of Chl a) that is only ~35% that of the expected value for LHCII trimers. It actually
FIGURE 4. Electron micrographs of proteoliposomes showing the size and the shape of the vesicles (black arrows). A, proteoliposomes containing LHCII. B, proteoliposomes containing reconstituted LHCBM1.

FIGURE 5. Single molecule fluorescence measurements. A, histogram of the estimated number of trimeric LHCII complexes associated with one bright spot found in confocal fluorescence images of surface-bound proteoliposomes. The red histogram represents a control experiment on trimeric LHCII complexes in detergent. The first peak at ~30 complexes of proteoliposomes with a high concentration of protein can be assigned to single proteoliposomes; larger values might also represent multiple liposomes within the confocal excitation spot. The blue histogram clearly shows the presence of monomeric LHCII complexes. The black dotted line indicates the hypothetical surface area of a sphere (diameter of 50 nm) purely made out of LHCII complexes. The inset illustrates a typical fluorescence image. arb., arbitrary units. B, exemplary fluorescence intensity traces of three LHCII proteoliposomes corresponding to the indicated protein concentration. The light and dark blue traces correspond to two different proteoliposomes from the same sample concentration of 0.1 μg of Chl a. The black dotted lines indicate exemplary intensity levels used for the lifetime analysis in D. C, the corresponding fluorescence decays of the fluorescence traces in Fig. 5B. The blue dotted line indicates a mono-exponential fit. D, the average lifetime corresponding to individual intensity levels for liposomes with the highest and lowest protein density. The slope in such a graph is an indication of the absorption cross and, therefore, clearly demonstrates the difference in the number of incorporated complexes. A smaller slope corresponds to a larger absorption cross-section as indicated by the red arrow. The dotted lines show the calculated slopes for an LHC monomer and LHCII trimer for comparison.
better matches the fluorescence intensity of monomeric LHC complexes than that of the trimer. The absorption cross-section distribution for that sample in Fig. 5A also clearly shows a peak at less than half a trimeric complex. Interestingly, most of the proteoliposomes with 0.1 μg of Chl a (~7000:1 lipid/protein molar ratio) show a fluorescence intensity equivalent to ~1:3 monomeric complexes. To further clarify this monomerization effect, we measured proteoliposomes that were loaded directly with monomeric LHCBM1 (~7000:1 lipid/protein molar ratio), and here we indeed found a matching fluorescence intensity between isolated monomeric complexes in detergent and single complexes in liposomes (see Fig. 6A). Finally, we confirmed the monomeric state of incorporated LHC complexes using LHCII from plants, the CD spectrum of which differs for monomers and trimers (28); LHCII trimers of tobacco inserted into liposomes show a CD spectrum typical for the monomeric state (Fig. 6B) (28).

A remaining question concerns the presence of free pigments in proteoliposomes. To that end, we recorded multiple fluorescence spectra (1-s integration time) of single proteoliposomes, which allows us to follow the spectral evolution over 30 s. The majority of measured proteoliposomes (LHCII; 7000:1 lipid/protein molar ratio) indeed exhibit small blue shoulder below 680 nm as illustrated in Fig. 7. The amplitude of that shoulder varies between different particles but always decreases exponentially in time, and only the main emission peak at 681 nm remains. Control experiments on liposomes that contained only free Chl (not shown) displayed the same exponential bleaching feature and had a fluorescence peak emission of 678 nm, matching the blue shoulder peak found in proteoliposomes.

Discussion
One of the main challenges in studying the mechanisms of photoprotection and the related regulatory components is the tradeoff between in vivo measurements, which are hard to interpret on a molecular scale, and in vitro experiments that usually fail to mimic the native and possibly essential thylakoid membrane environment. Proteoliposomes, where light-harvesting complexes are embedded in a lipid bilayer that contains the lipids of the thylakoid membrane, are becoming a popular system to study antenna complexes in vitro (20, 21, 24, 25, 27, 32, 33). With the aim to mimic the thylakoid membrane, highly concentrated proteoliposomes with a lipid/protein molar ratio between ~30–600:1 are used. Here we show that the aggregation feature in the fluorescence emission spectra at 77 K disappears only at a lipid/protein molar ratio of ~7000:1 (equal to 0.1 μg of Chl a in our liposome preparation), whereas lower lipid/protein ratios result in different levels of aggregations, which are characterized by different fluorescence lifetimes. It is thus important to closely monitor the lipid/protein ratio in the liposomes to avoid complications due to differences in the aggregation states of the samples.

By coupling bulk and single molecule measurements, we show that it is possible to control the amount of incorporated complexes from dozens of proteins down to a single complex per liposome, demonstrating that LHCII in a lipid environment, but in the absence of aggregation/clustering, is in an unquenched state. In fact, the fluorescence decay of most single complexes in the light harvesting state (“on-state” in blinking) can be fitted with a single lifetime component of ~3 ns, equivalent to that of detergent-isolated complexes. Even proteoliposomes that contain up to three or four monomeric complexes show this long lifetime component, as the complexes are apparently not connected and emit individually. This clearly demonstrates that the lipid environment and the interaction of the complexes with lipids do not per se lead to a quenched conformation. Our results instead suggest that protein-protein interactions and crowding are the main reasons for the shorter lifetime observed for LHCII in the membrane.

A somewhat surprising result is that trimeric LHCII disassemble to monomers upon incorporation into liposomes. This effect is unexpected as LHCII trimers are very stable even in detergent environment (34), and it has been shown that trimerization is favored by the presence of lipids (35). One possibility is that the surface curvature of liposomes, which corresponds to an angular change of ~10° within 5 nm, facilitates...
monomerization. Indeed, it has been shown that the curvature represents a stress that induces modification in some membrane proteins (36). Interestingly, such a degree of curvature is similar to that of the grana margin of the thylakoid membrane (37). These regions are particularly important for the regulation of association of LHCII with the two photosystems (38) and for the disassembling/repairing of photosystem II (39). It is thus tempting to speculate that the membrane curvature helps in the dissociation of multiprotein complexes. Indeed monomerization of LHCII has been observed in high light (40). Contrary to detergent micelles that effectively constrain and, therefore, facilitate the trimeric structure, reversible binding rates of monomeric subunits in a freely floating trimer might very well lead to the stochastically determined disassociation of the trimeric structure.

Finally, the fact that the blinking feature of single LHC complexes, well described for detergent-isolated complexes and assumed to be involved in NPQ (41), is confirmed in a lipid bilayer environment supports its functional relevance in native systems. Even proteoliposomes with a high protein density (black line in Fig. 5) show reversible intensity fluctuations that can be caused either by a few isolated complexes or more likely by switching events that affect a whole domain of connected complexes.

In conclusion, based on the controlled variation in lipid/protein ratio and the concomitant determination of the LHCII aggregation together with its spectroscopic properties, we conclude that LHCII has the tendency to cluster in the membrane, and we suggest that this clustering induces a conformational change of LHCII that controls the concentration of the excited states. We observe clustering already in the presence of a molar lipid to protein ratio of ~700:1, a condition in which LHCII is much more diluted than in the thylakoid membrane. This suggests that in vivo in its native membrane LHCII is normally clustered, which explains their shorter lifetime compared with isolated complexes.

Experimental Procedures

Strain, Growth Conditions, and Thylakoid Preparations—C. reinhardtii cells (strain JVD-1B[pGG1]) were grown in tris acetate phosphate liquid medium (42) at 25 °C under mild agitation (170 rpm) and 50 μmol of photons PAR m⁻² s⁻¹. Thylakoid membrane isolation was performed as described in Fischer et al. (43) with modifications reported in Drop et al. (44). Briefly, the cells were disrupted by sonication (60 watts power in 10 cycles of 10 s on/30 s off) and centrifuged at 17,000 g at 4 °C for 20 min. Purification of thylakoid membranes was performed using a discontinuous gradient in a SW41 swinging bucket rotor (24,000 rpm, 1 h, 4 °C).

Isolation of Light Harvesting Complexes—The isolation of LHCII was achieved following the protocol described in Drop et al. (30). In brief, unstacked thylakoids were solubilized in α-dodecylmaltoside detergent. The sample was fractionated by sucrose gradient density centrifugation. The green bands (LHCII) were harvested with a syringe.

In Vitro Reconstitution of Recombinant LHCII—The reconstitution of LHCBM1 was done as described in (45). In short, the apoprotein overexpressed in Escherichia coli and purified from inclusion bodies was mixed with pigments extracted from C. reinhardtii. The reconstituted protein was then purified by nickel affinity chromatography, taking advantage of a His tag on the C terminus of the protein and by sucrose density gradient centrifugation. The green bands (LHCII) were harvested with a syringe.

Liposome Preparation and Incorporation of LHCII—For liposome preparation we used a modified protocol from Gundermann and Büchel (46). The thylakoid lipids monogalactosyl diacylglycerol, digalactosyl diacylglycerol, sulfoquinovosyl diacylglycerols, phosphatidylglycerol, and phosphatidylcholine were purchased from Larodan Fine Chemicals (Malmö, Sweden), already dissolved in chloroform. Lipids were mixed in the
following relative molar ratios: 41.2% monogalactosyl diacylglycerol, 26.7% digalactosyl diacylglycerol, 15.6% sulfoquinovosyl diacylglycerols, 11.5% phosphatidylglycerol, and 5% phosphatidylcholine (26). The chloroform was slowly removed under N₂ vapor to form a thin film of lipids on the inner wall of the vial. 85 μg of lipids were rehydrated using 29 μl of 10% octyl glucoside solution and mixed overnight at 4 °C. Then 221 μl of 20 mm Tricine solution (pH 7.5) and 250 μl of dialysis buffer 4× (80 mm Tricine (pH 7.5), 20 mm MgCl₂, and 0.04% sodium azide) were added to the lipid solution. Extrusion was performed 10× through a 0.2-μm filter using the Extruder set (Avanti Polar Lipids, Inc.). The sample was diluted in a total volume of 125 μl dialysis buffer 1× and mixed with 125 μl of the lipid suspension. The final solution was then subjected to 3 cycles of vortexing (5 min) and incubation on ice (5 min). Finally, to remove the detergent and incorporate the protein into liposomes, the sample was dialyzed (14 kDa Viskin dialysis membrane) for 2 days in a dialysis buffer in ice. During the second day Biobeads-SM2 (Bio-Rad) were added to the dialysis buffer to remove the remaining detergent. The sample was then centrifuged for 30 min at 25,000 × g at 4 °C. The supernatant was collected and transferred in another tube.

Absorption, Circular Dichroism (CD), and Steady State Fluorescence Measurements—Room temperature absorption spectra were recorded with a Cary 4000 spectro-photometer (Varian). The CD spectra were measured using a Chirascan CD spectrophotometer (Applied-Photophysics). Fluorescence emission spectra were recorded using a Fluorolog 3.22 spectrofluorimeter (Jobin Yvon-Spex). The excitation wavelengths were 440, 475, and 500 nm, and emission was detected in the 600–800-nm range. Excitation and emission bandwidths were set to 3 nm. All fluorescence spectra were measured at an absorbance of 0.05 cm⁻¹ at the maximum of the Qy absorption band. Room temperature measurements were performed in 0.5 μM sucrose, 20 mM Hepes (pH 7.5), and 0.06% α-dodecylmaltoside buffer. For 77 K measurements a liquid nitrogen cooled device was used (cold finger).

Time-correlated Single Photon Counting—Time-resolved fluorescence measurements were performed using a FluTime 200 fluorometer (PicoQuant). The samples were diluted to an absorbance of 0.05 cm⁻¹ at the Qy maximum and stirred in a cuvette with a path length of 1 cm. Excitation was provided by a 470-nm laser diode with a repetition rate of 10 MHz and a power of 5 microwatts. Fluorescence emission at 680 nm was collected under an angle of 90° with respect to the excitation. All measurements were performed at 283 K, and the maximum number of counts in the peak channel was 20,000. Kinetics were analyzed after deconvolution from the instrument response function and measured with a pinacyanol iodide dye dissolved in methanol in which the lifetime is ~6 ps (47). Full width at half maximum of the instrument response function was ~80 ps. Global analysis was made using the TRFA Data Processor Advanced software.

Electron Microscopy—A 3-μl sample was applied on a holey carbon-coated grid (Quantifoil 3.5/1), blotted, and vitrified using a Vitrobot (FEI). Frozen hydrated samples were observed in an FEI Tecnai T20 electron microscope operating at 200 keV, equipped with a Gatan model 626 cryo-stage. Images were recorded on a slow-scan CCD camera under low dose conditions.

**Light-harvesting Complexes in Lipid Environment**

**Single Molecule Spectroscopy**—Proteoliposomes were immobilized on poly-L-lysine-coated microscope cover glasses, and the measuring solution (1× dialysis buffer) within the closed sample chamber contained an oxygen scavenging mix (2.5 mM protocatechuic acid, 25 mM protocatechuate-3,4-dioxigenase) to avoid photo-bleaching via singlet oxygen formation. The concentration of proteoliposomes was empirically adjusted to achieve a surface density of <10 particles per fluorescence image (5 × 5 μm). The rupture of vesicles upon nonspecific surface binding cannot be excluded, but surface coating with poly-L-lysine has been reported to prevent disruption of lipid vesicles on the bare glass interface and is a standard technique for cell and protein adhesion (48, 49). Single particles were excited at 633 nm with a repetition rate of 76 MHz (Coherent Mira 900F coupled to a Coherent MIRA OPO), and fluorescence spectra, intensities, and lifetimes were measured at room temperature on a home-built confocal fluorescence setup as described in detail earlier (50). In short, the fluorescence intensity was detected with avalanche photodiodes (Micro Photon Devices, PDM series, instrument response function of 38 ps or SPCM-AQR-16, PerkinElmer Optoelectronics, instrument response function of 304 ps), and fluorescence decay histograms were obtained via time-correlated single photon counting (PicoHarp 300, PicoQuant) and fitted with a deconvolution algorithm in FluoFit (PicoQuant). Fluorescence intensity traces were binned into 30-ns time steps, and the fluorescence decay traces were binned into 4-ps intervals. Fluorescence spectra with an integration time of 1 s were obtained by dispersing the collected fluorescence via a grating (HR830/800 nm, Optoelectronics LLC) onto a CCD camera (Spec10:100BR, Roper Scientific). The relative absorption cross-section of identified particles can be calculated as,

\[
\sigma = \frac{k_{abs}}{I_0/E_{photon}}
\]

(Eq. 1)

with

\[
k_{abs} = \frac{1}{\eta_{acq} \sum_{i} A_i \tau_i} = \frac{\sum A_i}{\eta_{acq} \tau_{bin} k_f}
\]

(Eq. 2)

where \(k_{abs}\) is the photon absorption rate, \(I_0\) is the excitation intensity, \(E_{photon}\) is the energy of a fluorescence photon (680 nm), \(\eta\) is the overall detection efficiency of the confocal setup, \(t_{acq}\) is the acquisition time, \(t_{bin}\) is the binning time of the fluorescence decay histogram, and \(A_i\) and \(\tau_i\) are the amplitudes and lifetimes of the fitted multi-exponential fluorescence decay.

**Author Contributions**—R. C. conceived the study. A. N. and J. M. G. conducted most of the experiments and analyzed the results. L. D. set up the liposome system. M. C. A. S. performed the electron microscopy. R. v. G. provided access to the single molecule spectroscopy setup. A. N., J. M. G., and R. C. wrote the paper. All authors approved the final version of the manuscript.

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