to be identical to material prepared directly from amine rotamer (ASI Applied Systems, Millersville, Maryland, USA); this equipment makes it possible to assay functional group changes as they occur.

Cl2C=O and Et3N.

Figure 5 Spectroscopic evidence that carbonyl dichloride fuels the unidirectional rotation of 2. Partial 1H NMR spectra (monitoring the bridgehead proton relative to (CH3)4Si=O p.p.m.) indicate the sequence of events as a function of time. Numbers next to peaks in the spectra refer to structures in Fig. 3. Numbers indicate the conversion of 5 to 6 being >80% complete in ~6 h. Urethane 6 was isolated and shown to be identical to material prepared directly from amine rotamer 7 by reaction with Cl2C=O/Et3N. Control experiments established that 6 does not convert to 5, that is, that the conversion of 5 → 6 is unidirectional. The assignment of structures to 3-6 is corroborated by infrared spectroscopic monitoring of the relevant reactions of 2, 7 and 9 with Cl2C=O/Et3N (and also 10 in the case of 8), performed using ReactIR technology (ASI Applied Systems, Millersville, Maryland, USA); this equipment makes it possible to record in situ, and in real time, the infrared spectra of reaction mixtures and to thereby assay functional group changes as they occur.

Light-driven monodirectional molecular rotor

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Attempts to fabricate mechanical devices on the molecular level have yielded analogues of rotors, gears, switches, shuttles, turnstiles and ratchets. Molecular motors, however, have not yet been made, even though they are common in biological systems. Rotary motion as such has been induced in interlocked systems and directly visualized for single molecules. This work was supported by the NIH.

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axial chirality and the presence of two chiral centres are essential for the observed monodirectional behaviour of the molecular motor. Two light-induced cis-trans isomerizations are each associated with a 180° rotation around the carbon-carbon double bond and are each followed by thermally controlled helicity inversions, which effectively block reverse rotation and thus ensure that the four individual steps add up to one full rotation in one direction only. As the energy barriers of the helicity inversion steps can be adjusted by structural modifications, chiral alkenes based on our system may find use as basic components for ‘molecular machinery’ driven by light.

The structure of the molecular rotor (3R,3'R)-(P,P)-trans-1,1',2',2',3',3',4',4'-octahydro-3,3'-dimethyl-4,4'-biphenanthrylidene (1 in Fig. 1) features two identical halves connected by a central carbon-carbon double bond. (Here P denotes the right-handed helicity in each half of the structure; M, used later, stands for left-handed helicity.) The design of the molecular rotor presented here is based on the exploitation of two fundamental principles. First, the light-induced trans to cis isomerization around a carbon-carbon double bond is an extremely fast and reversible process; in nature, it forms the basis for the information-retrieval step in the process of vision. Second, the concerted action of two chiral elements in a single chemical or physical event can lead to a unique handedness via the process of chiral discrimination, one of the essential features in living organisms. Steric interference in the molecule imposes a helical shape to the structure. Trans-1 and cis-2 are isomeric structures following a 180° rotation around the central bond, a process which is accomplished by irradiation with ultraviolet light at room temperature.

The configuration at the stereogenic centre, bearing the methyl-substituent, is (R) for both halves; Mtrans and Mcis (Fig. 1) are axial and equatorial oriented methyl-substituents, respectively. The synthesis of trans-(3R,3'R)-(P,P)-1 and the unequivocal determination of the absolute stereochemistry and molecular structures of trans-1 and cis-2 have been reported. We studied the photochemical switching behaviour of 1, and Fig. 1 shows the dynamic processes and the different stereoisomeric structures that were observed. The absorption spectra of the stereoisomers of 1 and 2 are given in Fig. 1.

Irradiation (wavelength \( \lambda \geq 280 \, \text{nm} \)) of (P,P)-trans-1 in n-hexane at -55 °C resulted in the formation of (M,M)-cis-2. A cis-2 to trans-1 ratio of 95:5 is observed at photoequilibrium. The isomerization is fully reversible, and the photoequilibrium shifts to (P,P)-trans-1 as the main isomer on irradiation with \( \lambda \geq 380 \, \text{nm} \). The trans-cis photoisomerization is evident from \(^{1}H\) NMR analysis, and the simultaneous reversal of P to M helicity is detected by circular dichroism (CD) (Fig. 3, traces A and B).

When the temperature of the solution of (M,M)-cis-2 is raised to 20 °C, a fast and selective conversion to (P,P)-cis-2 is observed and the concomitant change in CD absorption (Fig. 3, traces B and C) shows the helix reversal associated with this thermal interconversion. Irradiation of (P,P)-trans-1 under ambient conditions leads to complete conversion to (P,P)-cis-2, as the unstable (M,M)-cis-2 is removed from the photoequilibrium in a fast thermal isomerization. Although (P,P)-cis-2 was prepared by photochemical isomerization of (P,P)-trans-1 (ref. 16) we have now established that there is no direct pathway between these isomers. It should be emphasized that the (M,M)-cis-2 to (P,P)-cis-2 isomerization is irreversible under the thermal or photochemical conditions that were employed. Subsequent irradiation of (P,P)-cis-2 at \( \lambda \geq 280 \, \text{nm} \) results in the formation of (M,M)-trans-1, reaching a photoequilibrium with a ratio of (M,M)-trans-1 to (P,P)-cis-2 of 90:10. The change in CD absorption that accompanies this photochemical cis-trans isomerization (Fig. 3, traces C and D) is again associated with a thermal reversion for both the trans- and the cis-isomerization of (P,P)-trans-1 to (M,M)-trans-1 is not observed.

The main question is why the thermal M,M to P,P double helix reversal for both the trans-isomer 1 and the cis-isomer 2 are irreversible. Molecular modelling studies (MOPAC 93-AM1, Fujitsu, Tokyo, 1993) of (3R,3'R)-trans-1 show two stable forms, (P,P)-trans-1 and (M,M)-trans-1, with an energy difference (\( \Delta E \)) of 8.6 kcal mol\(^{-1} \) in favour of the (P,P)-isomer in accordance with the experimental observations. In the more stable (P,P)-isomer, the two methyl substituents are in axial positions (to prevent steric hindrance) whereas in the less stable (M,M)-isomer the methyl groups adopt equatorial orientations. For the cis-isomer 2 the two methyl groups are again in an axial orientation in the most stable (P,P)-cis-2 form.

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**Figure 1** Photochemical and thermal isomerization processes of (P,P)-trans-1. UV irradiation with high pressure Hg-lamp, Pyrex filter, \( \lambda \geq 280 \, \text{nm} \) or Xe-lamp, Toshiba L-3g glass filter, \( \lambda \geq 380 \, \text{nm} \). First order kinetics were observed for the thermal processes and temperature dependent \(^{1}H\) NMR and CD studies in the range 50.0-81.1 °C gave \( \text{Ea} = 26.4 \, \text{kcal mol}^{-1} \) for the (M,M)-trans-1 to (P,P)-trans-1 interconversion. It should be noted that no racemization takes place during any of the photochemical or thermal steps as was proven by chiral HPLC analysis of the isomers obtained after the individual steps.

**Figure 2** Ultraviolet-visible spectra (hexane solvent). Trace A, (P,P)-trans-1; trace B, (M,M)-cis-2; trace C, (P,P)-cis-2; trace D, (M,M)-trans-1.
whereas in the less stable \((M,M)\text{-cis-2}\) form \((\Delta E = 11.0 \text{ kcal mol}^{-1})\) the two methyl groups take equatorial positions.

The above results indicate that irradiation of \((P,P)\text{-trans-1}\) at \(\lambda \geq 280 \text{ nm}\) at 60 °C will result in the formation of all three other states of the four-step isomerization cycle (shown in Fig. 1) consecutively, because under these conditions both photoisomerization steps as well as the two thermal isomerization steps will take place. The inset of Fig. 3 shows the change in CD signal at 217 nm, as monitored during three full cycles. When one considers the structural changes observed during the full cycle shown in Fig. 1 it is evident that, relative to the bottom half of the molecule (considered the static part), the top half of the molecule (considered the rotor) undergoes a full 360° rotation exclusively in a clockwise sense (as seen from the side of the static part). To the best of our knowledge, this is the first observation of a process in which inversion of a chiral centre occurs, and the methyl groups adopt the more favourable axial orientation. Next, a second photochemical step converts \(\text{cis-2}\) to \(\text{trans-1}\) with simultaneous \(P,P\) helicity inversion; but in this step again, as a consequence of the rotation around the alkene double bond, the methyl groups adopt the less favourable equatorial orientation. Finally, \((M,M)\text{-trans-1}\) relaxes to the starting isomer \((P,P)\text{-trans-1}\) with simultaneous \(M,M\) to \(P,P\) helix inversion and a change of the orientational helicity occurs, and the fixed \((3R,3'R)\text{-cis}\) enantiomers; \((P,P)\) or \((M,M)\) helicity; the fixed \((3R,3'R)\text{-configuration at the sterogenic centres; conformational flexibility of the cycloalkene rings; and the axial or equatorial orientation that can be adopted by the methyl groups.}

The stereochemical changes during the entire rotation cycle that dictates the clockwise monodirectional behaviour for \((P,P)\text{-trans-1}\) can be rationalized as follows. Photochemical conversion of \((P,P)\text{-trans-1}\) inverts the \(P,P\) helicity to \(M,M\) helicity. Simultaneously, the axial methyl substituents in \((P,P)\text{-trans-1}\) adopt the less favourable equatorial orientation in \((M,M)\text{-cis-2}\). The photochemically induced energetically uphill-process is followed by a thermal downhill process in which inversion of \(M,M\) to \(P,P\) helicity occurs, and the methyl substituents adopt the more favourable axial orientation. Subsequently, two energetically downhill processes generate the more stable \((P,P)\text{-cis-2}\) and \((P,P)\text{-trans-1}\) isomers. As the helicity inverts simultaneously in the last two steps, the reverse rotation pathway is effectively blocked. Starting with \((3R,3'R)\text{-cis}\), clockwise rotation is therefore found; if the \((3S,3'S)\text{-enantiomer of this molecular rotor were used, anti-clockwise rotation is expected.}

The cycle shown in Fig. 1 may be regarded as four distinct states in an optical molecular switch, each of which can be populated depending upon temperature and wavelength of the light used; at the appropriate wavelength and temperature, a continuous monodirectional rotation is induced in the system. The origin of this rotation process lies in the molecular architecture of structure 1, where the following stereochemical features are all important: thermally stable, but photochemically interconvertable \(cis\) and \(trans\) isomers; \((P,P)\) or \((M,M)\) helicity; the \((3R,3'R)\text{-configuration at the stereogenic centres; conformational flexibility of the cycloalkene rings; and the axial or equatorial orientation that can be adopted by the methyl groups.}

Lek mating systems are characterized by males displaying in groups. The main benefit from group display is thought to be an increase in the number of females arriving per male. However, when mating success is highly skewed it is not clear why unsuccessful males participate in group display. In theory, all males on leks could obtain indirect fitness benefits if displaying groups consisted of related individuals. Here we present two independent sets of data that show that peacocks (*Pavo cristatus*) display close to their kin. DNA fingerprinting showed that males at Whipsnade Park were more closely related to males within the same lek than to males at other leks. Separately, we found that after an experimental release of a mixed group of related and unrelated males, brothers (paternal sibs or half-sibs) established permanent display sites very close together. This result is unexpected, as the released birds could not become familiar with their brothers during their development. The released young were hatched from eggs that had been removed from their parents shortly after laying and mixed with the eggs of non-relatives. These data indicate that birds can evolve a means of kin association that does not involve learning the characteristics of relatives or the use of environmental cues. If social learning is not necessary for kin association then kin effects may be of more widespread importance in avian social interactions, and in particular in the evolution of lek mating, than previously appreciated.

Males in lek mating systems aggregate to display to attract females. Group display has been shown to increase the number of females arriving per male. We used multilocus fingerprinting to compare the genetic similarity (measured as the degree of band-sharing) within and between these lek sites. The degree of band-sharing within leks was significantly higher than that between leks (within leks mean $S = 0.816$, $n = 4, 6, 4$ and $7$, respectively) were significantly different ($P = 0.01$; Mantel randomization test, 100,000 randomizations, $P = 0.01$; Fig. 2). Assuming that birds in different leks are unrelated and the detected minisatellites segregate independently, the increased band-sharing within leks is close to that expected for half-siblings ($0.810$).

How could peacocks come to display near to their close kin? Peacocks take no part in reproduction after mating and therefore young birds cannot learn the identity of their fathers. One possibility that could result in a tendency for related birds to display...