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Enhanced selectivity in non-heme iron catalysed oxidation of alkanes with peracids: evidence for involvement of Fe(IV)=O species.

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Supporting information

Figure S1 UV-Vis spectrum before (red), during (green) and after (blue) catalytic oxidation of cyclohexane in CH₃CN with *m*CPBA/**1a**, under N₂.

Figure S2 UV-Vis spectrum before (red), during (blue) and after (green) catalytic oxidation of cyclohexane in CH_3CN with *m*CPBA/**1a**, under O_2 .

Figure S3 UV-Vis spectrum before (red) and 24 h after (reaction) catalytic oxidation of cyclohexane in CH_3CN with *m*CPBA/**1a**, under O_2 and after bulk reduction at -0.2 V vs SCE (see legend figure S4) (black).

Figure S4 Cyclic voltammogram before (red) and 24 h after reaction (green) catalytic oxidation of cyclohexane in CH_3CN with *m*CPBA/**1a**, under O_2 and after bulk reduction (black) at -0.2 V vs SCE (Working electrode Pt Gauze, Counter electrode Pt Wire, using CHInstruments CHI630B). Initial scanning point and direction indicated by black arrow.

Analytical Data.

1a and **1b** were prepared analogous to literature procedures. Spectroscopic data are consistent with the structures.ⁱ Anal. calcd. for **1a**, C₂₅H₃₀B₂F₈FeN₅·H₂O: C 45.77, H 4.00, N 12.81; found: C 45.95, H 3.88, N 12.57. Anal. calcd. for **1b** C₈₇H₄₅B₂F₄₈FeN₅ C 48.61, H 2.11, N 3.26; found: C 47.80, H 2.29, N 3.56.

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